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Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(O-B) = 0.004 \text{ Å}$ R factor = 0.017 wR factor = 0.029 Data-to-parameter ratio = 12.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Aragonite-type lanthanum orthoborate, LaBO₃

The crystal structure of the low-temperature (LT) modification of LaBO₃ has been redetermined from single-crystal X-ray data; the resulting structure confirms the previous study [Abdullaev, Dzhafarov & Mamedov (1976). *Azerbaidzhanskii Khim. Zh.* pp. 117–120], but with improved precision. LT-LaBO₃ crystallizes in space group *Pnma* and adopts the aragonite-type structure. Except for one O atom, which is situated on a general position, all other atoms (one La, one B and a second O atom) lie on mirror planes. The structure is composed of LaO₉ polyhedra with an average La–O distance of 2.593 Å and trigonal BO₃ groups with an average B–O distance of 1.373 Å. Slight anisotropies of the thermal vibrations of La and B atoms suggest that the electrostatic La…La and La…B interactions across the shared edges are weak.

Comment

The crystal structure of the aragonite-type low-temperature (LT) form of LaBO₃ was first determined by Abdullaev *et al.* (1976) in space group *Pnma*. In this structure determination, the structure was refined with isotropic displacement parameters and the reliability index converged to R(F) = 0.11. Recently, the deviation from the normal aragonite-type structure (space group *Pnma*) was reported for LT-LaBO₃, but no information on its structural parameters was provided (Giesber *et al.*, 2003). Thus, this structure has not been refined satisfactorily to date. An accurate structure model for this compound is indispensable because of its importance as the parent structure of Sr-doped LaBO₃ which is a promising high-temperature proton conductor (Shimura *et al.*, 2001; Amezawa



Figure 1

Representation of the aragonite-type $LT-LaBO_3$ structure. LaO_9 groups are shown as hatched polyhedra and B atoms as black spheres. B-O bonds are indicated as solid black lines.

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Figure 2

Part of the aragonite-type LT-LaBO₃ structure with displacement ellipsoids drawn at the 65% probability level. [Symmetry codes: (i) $\frac{1}{2} + x, y, \frac{1}{2} - z;$ (ii) $-\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} + z;$ (iii) $-\frac{3}{2} - x, -\frac{1}{2} + y, \frac{1}{2} + z;$ (iv) $x, -\frac{3}{2} - y, z;$ (v) $-2 - x, \frac{1}{2} + y, 1 - z;$ (vi) $-\frac{3}{2} - x, -2 - y, \frac{1}{2} + z;$ (vii) -2 - x, -2 - y, 1 - z.]

et al., 2004). We report here a full structure refinement of the aragonite-type LT-LaBO₃, including an anisotropic model of the thermal vibration.

Giesber *et al.* (2003) reported that the aragonite-type LaBO₃ belongs to the non-centrosymmetric space group $Cmc2_1$. However, from systematic absences and intensity statistics based on Wilson analysis, we conclude that the most probable space group of this compound is the centrosymmetric space group *Pnma*, as reported by Abdullaev *et al.* (1976). In the present study, the crystal structure was satisfactorily refined in the normal aragonite-type structure, and bond valence sums (Brown & Altermatt, 1985) resulted in reasonable values [2.98 valence units (v.u.) for La, 2.98 v.u. for B, 1.91 v.u. for O1 and 2.02 v.u. for O2].

As shown in Fig. 1 and Table 1, the La atom is coordinated by nine O atoms with a mean bond length of 2.593 Å, and the B atom is planar-triangularly coordinated by three O atoms with a mean bond length of 1.373 Å. Each LaO₉ polyhedron shares edges with six other LaO₉ polyhedra and three BO₃ triangles, whereas each BO₃ triangle shares edges with three LaO₉ polyhedra.

In compounds with shared edges between coordination polyhedra, thermal vibrations of cations can be restricted in the cation-cation directions across the shared edges because of electrostatic repulsions between these cations, as has been observed in garnet compounds (Nakatsuka *et al.*, 1995, 1999, 2003, 2004*a*,*b*; Yoshiasa *et al.*, 1997). However, the displacement ellipsoids of the La and B atoms in the present crystal structure indicate only a slight anisotropy (Fig. 2), suggesting that the electrostatic La···La and La···B interactions across the shared edges are weak.

Experimental

In a recent study we reported on the crystal structure of cubic perovskite-type single crystals of LaAlO₃, synthesized at 4.5 GPa and 1273 K (Nakatsuka *et al.*, 2005). During this high-pressure synthesis, single crystals of the aragonite-type LT-LaBO₃ were obtained as a

minor by-product due to contamination by boron from the boron nitride capsule used in the high-pressure cell assembly. The single crystal of LT-LaBO₃ used in the present study was selected from these products. The detailed procedure of the high-pressure synthesis is described elsewhere (Nakatsuka *et al.*, 2005).

Crystal data

LaBO₃ $M_r = 197.71$ Orthorhombic, *Pnma* a = 5.8744 (3) Å b = 5.1087 (3) Å c = 8.2581 (5) Å V = 247.83 (2) Å³ Z = 4 $D_c = 5.299$ Mg m⁻³

Data collection

Rigaku AFC-7R diffractometer ω -2 θ scans Absorption correction: part of the refinement model (ΔF) (Walker & Stuart, 1983) $T_{min} = 0.418, T_{max} = 0.509$ 1591 measured reflections 396 independent reflections 369 reflections with $F^2 > 2\sigma(F^2)$

Refinement

Refinement on *F* $R[F^2 > 2\sigma(F^2)] = 0.017$ $wR(F^2) = 0.029$ S = 1.98369 reflections 29 parameters $w = 1/[\sigma^2(F_o) + 0.00014|F_o|^2]$ Mo $K\alpha$ radiation Cell parameters from 50 reflections $\theta = 22.5-25.0^{\circ}$ $\mu = 16.88 \text{ mm}^{-1}$ T = 296 KIrregular fragment, colorless $0.06 \times 0.05 \times 0.04 \text{ mm}$

$$\begin{split} R_{\rm int} &= 0.030 \\ \theta_{\rm max} &= 30.0^{\circ} \\ h &= -8 \rightarrow 8 \\ k &= 0 \rightarrow 7 \\ l &= -11 \rightarrow 11 \\ \text{3 standard reflections} \\ \text{every 150 reflections} \\ \text{intensity decay: 0.6\%} \end{split}$$

 $\begin{array}{l} (\Delta/\sigma)_{max}=0.001\\ \Delta\rho_{max}=1.24 \ e \ A^{-3}\\ \Delta\rho_{min}=-0.98 \ e \ A^{-3}\\ \text{Extinction correction: Zachariasen}\\ (1967) \ type \ 2 \ Gaussian \ isotropic\\ \text{Extinction coefficient: } 0.056 \ (3) \end{array}$

Table 1

Selected geometric parameters (Å, $^{\circ}$).

La-O1 ^{viii}	2.451 (3)	La-O2 ^{xiii}	2.629 (2)
La–O1 ^{ix}	2.723 (1)	La-O2 ^{xiv}	2.601 (2)
La–O1 ^x	2.723 (1)	La-O2 ^{xv}	2.601 (2)
La-O2 ^{ix}	2.491 (2)	B-O1 ⁱ	1.374 (5)
La–O2 ^{xi}	2.491 (2)	B-O2	1.373 (3)
La–O2 ^{xii}	2.629 (2)	B-O2 ^{xvi}	1.373 (3)
$O1^{viii} - B - O2$	120.3 (2)	O2-B-O2 ^{xvi}	119.1 (3)
$O1^{viii} - B - O2^{xvi}$	120.3 (2)		()
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Symmetry codes: (i) $x + \frac{1}{2}, y, -z + \frac{1}{2}$; (vii) $x + \frac{1}{2}, -y + \frac{1}{2}, -z + \frac{1}{2}$; (ix) $-x + \frac{1}{2}, -y, z + \frac{1}{2}$; (x) $-x + \frac{1}{2}, -y + 1, z + \frac{1}{2}$; (xi) $-x + \frac{1}{2}, y + \frac{1}{2}, z + \frac{1}{2}$; (xii) $x - \frac{1}{2}, -y + \frac{1}{2}, -z + \frac{1}{2}$; (xiii) $x - \frac{1}{2}, y, -z + \frac{1}{2}$; (xiv) $-x + 1, y + \frac{1}{2}, -z + 1$; (xv) -x + 1, -y, -z + 1; (xvi) $x, -y + \frac{1}{2}, z$.

The present study confirms the basic structural features determined from the previous investigation (Abdullaev *et al.*, 1976) with a much higher precision by about a factor of 10 (note that in the previous study the space-group setting *Pmcn* was chosen). The highest peak and the deepest hole in the final difference Fourier map are located 0.77 and 1.17 Å, respectively, from atom O1.

Data collection: *WinAFC* (Rigaku Corporation, 1999); cell refinement: *WinAFC*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1999); program(s) used to solve structure: *TEXSAN*; program(s) used to refine structure: *TEXSAN*; molecular graphics: *ATOMS* (Dowty, 2000); software used to prepare material for publication: *TEXSAN*.

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